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Original Article

Chemical Profiling of Minerals and Elemental Composition of Rock and Soil Samples from Bakin Dutse Hills, Madagali, Adamawa State Nigeria

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A B S T R A C T

Evaluation of natural resources via chemical profiling is a prerequisite to mining and material sourcing, which are important activities in positioning an economy for the future. This study explores the untapped rocks located within the Bakin Dutse hills in Adamawa State, northeastern Nigeria. The minerals and elemental compositions of the rock samples were determined using X-ray diffractometer (XRD) and X-ray fluorescence (XRF) instruments respectively. Results of the mineral composition of the rock samples showed (in order of abundance) the presence of quartz, albite, annite, and microcline, as well as orthoclase, phlogopite, oligoclase, chrysotile, and cordierite which were sparsely found in the study area. The abundant presence of silicate and aluminum silicate minerals in the study area was also shown in the major elemental composition of the rock samples, which follow the order SiO₂ (55.76 - 63.81%) > Al_2O_3 (11.06 - 13.15%) > Fe_2O_3 (8.41 -11.2% > K₂O (6.12 - 7.75%) > CaO (2.29 - 6.37%) > Na₂O (2.4%) > MgO (0.09 -2.2%) > SO₃ (0.37 – 1.05%). The minor elements in the rocks include PbO, TiO₂, BaO, CuO, ZnO, P2O5, ZrO2, Rb2O, MnO, and Cl. The mineral and elemental compositions of the soil in the study area show similarities that link the origin of the soil to the weathering of the rocks. The geo-chemical data provided in this study will make an important contribution to both the literature and the database for future mining prospects in the study area.

Introduction

ocks are combinations of solid minerals, which are natural components of the environment. Solid minerals are natural inorganic materials with a specific range of chemical and physical characteristics [1, 2]. Beyond industrial and material applications, components of rocks may also constitute green and sustainable energy potentials, which may require comprehensive evaluation and exploration for adequate conservation and utilization [3-6].

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Therefore, the chemical composition of rocks can be used as the basis for studying their economic potential in different fields of application and valuation. Soil can be basically described as disintegrated rocks in disproportionate mixture with organic matter, water, air, biotic, and abiotic components that may be capable of supporting plant life [7-9].

The economic potential of Africa still lies deeply in its ability to maximize the exploration and profits of the vast variety of minerals that are naturally deposited in countries within the boundaries of the continent. It has been estimated that about 90%, 81%, 62%, 70%, 50%, and 30% of the global deposits of diamond, cobalt, platinum, gold, chromium, and copper respectively are seated in Africa [10]. Nigeria is an economic giant in Africa, and the sources of wealth of the nation remain majorly tied to natural resources [11-13].

Nigeria has continued to rank high among the ten largest crude oil producers in the world [14]. However, before the discovery of oil in Nigeria in 1964, exploration and mining of solid minerals such as coal, tin, columbite, lead, gold, and zinc were on a commercial scale [15,16].

The Geological Survey of Nigeria, which was established in 1919, was saddled with the continued survey of solid mineral deposits in the country [17,18]. Nevertheless, the geochemical data for a significant portion of Nigeria is yet to be established [19].

According to Vision 2020 of the National Technical Group working on minerals and metals development, as re-presented by Falade and Adeyeye, [17], the entire northeastern Nigeria, which includes Adamawa State, is less dense with mineral deposits. However, this report may not be substantive because the geochemical data of a vast proportion of northeastern Nigeria is yet to be adequately studied and established.

Evaluation of geochemical data of the rocks and soil in an area may unlock the resource potential of the area. The knowledge of elemental composition in rocks may provide important information about the minerals that are abundant in the core of the rock [20,21]. Potential maps for rock minerals in the Southern New England Orogen, New South Wales, and Australia were created based on mineral model systems and high-quality geochemical data [22].

The mineral system models and geochemical data were summarily evaluated by: (1) a spatial data table detailing the translation of the mineral model system, (2) the predictive maps that capture the mappable components of the targeted mineral system, and (3) the final mineral potential maps to develop the mineral system atlas [22]. Solid minerals are of vast industrial profitability [23].

Therefore, survey, development, and the final exploration of solid minerals are significant steps in the economic diversification and progression of a nation. Solid mineral deposits also vary over a very wide range of useful varieties. Evaluation and monitoring of chemical components of rock in the environment may also be used to monitor the effect of natural and anthropogenic activities on the chemical content of the rocks [7,24-27].

The lichen-mineral system was modeled to identify the fundamental interactions between microorganisms and solid minerals in rock [7]. It was further reported that fungi and other microorganisms involved in the interaction induced mineral disaggregation, hydration, dissolution, and secondary mineral formation, directly or indirectly, due to the release of organic acids and dissolved metabolic byproducts from the microorganisms [7]. The effects of field-induced changes in the pH of the rock reservoir environment on the chemistry of rock minerals were studied by Mohammed et al. [27]. Calcite, feldspar, barite, dolomite, and quartz, were studied in the research, and it was concluded that the changes in pH of the rock's reservoir environment alter the minerals wettability by controlling the development of charges on the minerals surface via mineral dissolution, ionic specie adsorption, and doublelayer compression [27]. This therefore necessitates the progressive search and

evaluation of all potential deposits of solid minerals.

About two decades ago, independent research focused on the determination of mineral composition as well as the associated major (> 1%) and minor (\leq 1%) elemental composition of rocks and soil in Adamawa State and other northeastern regions have opened a new chapter in the establishment of the geochemical data of the region [9,19,21,28-30]. Following this trend in our previous studies, we have reported the minerals and elemental composition of rocks at Ditera and Waltadi, in the Song local government area, and the Sukur Hills in the Madagali local government area [23,31-32].

In this study, we evaluated the mineral and elemental oxide composition of top rock and soil from Bakin Dutse Hills, Madagali, Adamawa State, Nigeria. The mineral composition and the elemental oxides of the rock samples were analyzed using X-ray diffraction (XRD) and X-ray fluorescence (XRF) instruments, respectively. This study is also in furtherance of our previous studies [23,31-32]. The research was undertaken because no study was found (to the best of our knowledge) on the rocks and soil within Bakin Dutse Hills, Madagali, Adamawa State, Nigeria. The results will therefore make tangible contributions to the geochemical information of the area.

Materials and Methods

Basic tools such as an angler's tool, hammer, chisel, soil thief, steel pestle and mortar, spatula, and platinum crucibles were used in obtaining samples and in the sample preparations. Acid (H_2SO_4 and HCl) and base (NaOH and NH₃OH) used were products of MERCK Chemicals Limited. X-ray diffraction and X-ray fluorescence spectrophotometer instruments were used for the mineral and elemental analysis, respectively.

Study Area

Bakin dutse rocks lay within latitude 10° 51"N and longitude13° 47" E. Bakin Dutse is located near Gulak in the Madagali local government area of Adamawa State, Nigeria. A topographical map of the area is displayed in Figure1.



Figure 1 Topographical map of Barkin Dutse rocks

S/N	Sample locations	Sample code names
1	Bakin dutse East Rock	BER
2	Bakin dutse North Rock	BNR
3	Bakin dutse South Rock	BSR
4	Bakin dutse Top-center Rock	BTR
5	Bakin dutse West Rock	BWR
6	Bakin dutse East Soil	BES
7	Bakin dutse North Soil	BNS
8	Bakin dutse South Soil	BSS
9	Bakin dutse Top-center Soil	BTS
10	Bakin dutse West Soil	BWS

Sampling Locations

Rock and soil samples were collected from five locations randomly selected on and around the Bakin Dutse rocks. Sampling locations were marked to adequately represent a quantitative top sample of the entire area. The samples were designated according to their location, as presented in Table 1.

Sample Collection and Treatment

Rock samples were collected according to methods reported in our previous study [23]. About 1 cm of the rock surface was scraped before samples were chiseled to a depth of about 10 cm. Soil samples were collected from soil around the same location as the rocks. Soil samples were taken at about 30 cm depth using a soil thief. Samples were labeled at the points of collection and transferred to the laboratory in clean polyethylene bags. Samples were sequentially watched with dilute acids (0.01 M H₂SO₄ and 0.01 M HCl), base (0.05 M NaOH and 0.05 M NH₃OH), and then three times with distilled water prior to drying under the sun for 12 hours. After drying, samples were crushed and grinded using a steel pestle and mortar and further sieved through a 10 μ m mesh to obtain fine powder with particles \leq 10 µm. The powdered samples were packed in 25 Gm bottles and labeled for instrumental analysis.

Analyses of Rock and Soil Samples

The mineral compositions of the powdered samples were analyzed using a Rayon X-ray diffraction (XRD) spectrometer (PANanalytical Netherlands). The elemental Empyrean,

composition of samples was evaluated using Xray fluorescence (XRF) spectrometer analysis (Energy Dispersive X-Ray Fluorescence spectrometer miniPal 4 PANanaltical). The conditions of instrumental analyses are the same as reported in the previous study [23]. For XRD, 0.35 g of powdered sample was introduced into the sample container of the XRD instrument. The X-ray light wavelength was 1.5 Å, and the diffraction angles (2θ) were recorded between 0.5 and 100°. Minerals were identified via instruments in-built computation (according to Braggs equation $n \lambda = 2dSin\theta$) and library references (d-spacing). For XRF, 2.0 g of powdered sample was mixed with 0.4 g of stearic acid as a binder for the sample pellets mounted onto the instrument for the scans.

Results and Discussion

XRD spectra of rock samples from the study area are depicted in Figure 2. Nine minerals were identified in the rock samples from the study area. Quartz was most abundant in the area, as it was found in all five study locations, with a percentage range of 21% (BNR)-39% (BER). The abundance of quartz has been consistently reported in the rocks studied in the northeastern part of Nigeria, and this study remains consistent with the reported trend [19,21,23,28-29,31-32].



Figure 2 X-ray diffraction spectra of (a) BER, (b) BNR, (c) BSR, (d) BTR, and (e) BWR

Location	Minerals_identified	Compound_name	Chemical_formula	Density g/cm ³	Crystal system	%composition
BER	Albite	Sodium Aluminium Silicate	Na1.96 Ca0.04 Si5.96 Al 2.04 O16.00	2.62	Anorthic	35.0
	Orthoclase	Potassium Aluminium Silicate	K4.00 Al4.00 Si12.00 O16.00	2.56	Monoclinic	28.0
	Quartz	Silicon Oxide	Si3.00 O6.00	2.65	Hexagonal	21.0
	Phlogopite	Potassium Magnesium Aluminium Silicate Hydroxide	K1.66 Na0.14 Ba0.10 Mg4.39 Fe1.61 Si5.34 Al2.66 O23.66.F0.34 H3.11	2.96	Monoclinic	16.0
BNR	Oligoclase	Sodium Calcium Aluminium Silicate	Na1.45 Ca0.55 Al2.55 Si5.45 O16.00	2.65	Anorthic	31.0
	Quartz	Silicon Oxide	Si 3.00 O6.00	2.26	Hexagonal	39.0
	Microcline	Potassium Aluminium Silicate	K1.90 Na0.10 Al2.00Si6.00O16.00	2.56	Anorthic	24.0
	Annite	Potassium Iron Aluminium Silicate	Si5.68 Al4.26 Fe3.88 K1.80 O24.00	3.06	Monoclinic	10.0
	Chrysotile	Magnesium Silicate	Mg12.00 Si8.00 O36.00	2.51	Monoclinic	6.0
BSR	Quartz	Silicon Oxide	Si 3.0 O6.00	2.65	Hexagonal	29.0
	Microcline	Potassium Aluminium Silicate	K1.9 Na0.10 Al 2.0 Si6 O16.0	2.56	Anorthic	26.0
	Albite	Sodium Aluminium Silicate	Na2.0 Al2.0 Si6.0 O16.0	2.64	Anorthic	25.0
	Annite	Potassium Magnesium Iron Aluminium Silicate	K _{1.86} Na _{0.14} Mg _{2.04} Fe _{2.82} Ti _{0.92} Al _{2.56} Si _{5.44} O _{24.0}	3.16	Monoclinic	20.0
BTR	Quartz	Silicon Oxide	Si3.0 O6.0	2.65	Hexagonal	37.6
	Albite	Sodium Aluminium Silicate	Na2.0 Al2.0 Si6.0 O16.0	2.64	Anorthic	29.7
	Microcline	Potassium Aluminium Silicate	K1.9 Na0.1 Al2.0 Si6.0 O16.0	2.56	Anorthic	20.8
	Annite	Potassium Iron Aluminium Silicate	Si4.64 Al6.31 Fe3.25 K1.36 O24.0	3.10	Monoclinic	6.9
	Cordierite	Magnesium Aluminium Silicon	Mg _{3.0} Al _{16.00} Si _{2.0} O _{72.0} Bi _{0.46}	2.61	Orthorhom bic	5.0
BW R	Albite	Sodium Aluminium Silicate	Na1.96 Ca0.04 Si5.96 Al2.04 O16.0	2.62	Anorthic	34.3
	Quartz	Silicon Oxide	Si _{3.0} O _{6.0}	2.65	Hexagonal	32.3
	Microcline	Potassium Aluminium Silicate	K1.90 Na0.10 Al2.0 Si6.0 O16.0	2.56	Anorthic	24.2
	Annite	Potassium Magnesium Iron Aluminium Silicate	K1.86 Na0.14 Mg2.04 Fe2.86 Ti0.92 Al2.56 Si5.44 O24.0	3.18	Monoclinic	09.1 1

Table 2 Characteristics of minerals identified in Bakin Butse rock samples

To basically evaluate the mineral composition of the soil in comparison with the rock at the location, soil samples from the rock locations were also analyzed. Figure 3 demonstrates the XRD spectra of the soil samples analyzed. The results showed that quartz maintained dominance, as it was found in all soil samples. Albite, annite, and microcline also reoccur in some of the soil samples from the study locations. However, there was a marked increase in the occurrence of phlogopite (found in four out of the five locations studied) and the emergence of kaolinite (which was not found in any of the rock samples) compared to the rock samples. The result showed that the soil in the study area may have been formed significantly from the weathering and metamorphosis of the rocks. Phlogopite has been well reported to be formed from the metamorphosis of rock minerals [36,38]. Kaolinite has also been reported to be formed from the hydrothermal transformation of aluminosilicate minerals such as orthoclase, oligoclase, chrysotile, and cordierite, which were found in some locations of the rocks but not in any soil sample [39-40]. Therefore, the presence of phlogopite, and







Loca	identified	Compound_name	Chemical formula	Density g/cm ³	Crystal system	%composition
BES	Quartz	Silicon Oxide	Si _{3.0} O _{6.0}	2.65	Hexagonal	91.0
	Annite	Potassium Iron Aluminium Silicate	Si5.10 Al6.9 Fe6.0 K1.98 Na0.02 O24.0	3.69	Monoclinic	9.0
BNS	Quartz	Silicon Oxide	Si _{3.0} O _{6.0}	2.64	Hexagonal	43.0
	Albite	Sodium Aluminium Silicate	Na1.96 Ca0.04 Si5.96 Al2.04 O16.0	2.62	Anorthic	29.0
	Phlogopite	Potassium Magnesium Iron Flora Aluminium Silicate	K1.84 Na0.08 Mg3.12 Fe2.34 Ti0.29 Mn0.12 Si5.94 AL2.01 O22.12 F1.88	3.12	Monoclinic	15.0
	Kaolinite	Aluminium Silicate	Al8.0 Si8.0 O36.00	2.57	Monoclinic	13.0
BSS	Quartz	Silicon Oxide	Si3.0 O6.00	2.64	Hexagonal	44.0
	Kaolinite	Aluminium Silicate Hydroxide	Al _{2.0} Si _{2.0} O _{9.00} H _{4.00}	2.61	Anorthic	36.0
	Phlogopite	Potassium Magnesium Titanium Iron Aluminium Silicate	K3.56 Na0.28 Mg6.0 Fe4.44 Ti1.56 Al4.48 Si11.52 O48.00	3.07	Monoclinic	20.0
BTS	Microcline	Potassium Aluminium Silicate	K _{1.90} Na _{0.10} Al _{2.0} Si _{6.00} O _{16.00}	2.56	Anorthic	41.6
	Quartz	Silicon Oxide	Si3.00 O6.00	2.65	Hexagonal	28.7
	Albite	Sodium Aluminium Silicate	Na _{1.96} C _{0.04} Si _{5.96} Al _{2.04} O _{16.00}	2.62	Anorthic	20.8
	Kaolinite	Aluminium Silicate	Al2.00 Si2.00 O9.00	2.56	Anorthic	6.9

Table 3 Characteristics of minerals identified in Ba	акіп в	utse soll	sample
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BWS	Phlogopite	Potassium Nickel Aluminium Silicate Hydroxide Silicon Oxide	Al _{2.16} Si _{5.84} Ni _{6.00} K _{20.00} O _{240.0} H _{4.00}	3.50	Monoclinic Hexagonal	2.0
2110	Quuitz	Sincon onlac	513.00 0 0.00	2.01	menagonai	1710
	Albite	Sodium Aluminium Silicate	Na1.96 Ca0.04 Si5.96 Al2.04 O16.00	2.62	Anorthic	43.4
	Phlogopite	Potassium Magnesium Titanium Iron Aluminium Silicate	K3.72 Na0.28 Mg6.00 Fe4.44 Ti1.56 Al4.08 Si11.02 O48.00	3.09	Monoclinic	9.1

kaolinite in the soil can be attributed to the weathering and metamorphic processes of the rocks in the area. Table 3 complementarily lists the identified minerals, their elemental composition, and the characteristic crystal lattices.

The results of the elemental components of the rock samples presented as major (more than 1% of the rock) and minor (within 1 to 0.1% of the rock) elemental components are presented in Tables 4 and 5, respectively. The elemental compositions which are presented in their oxide forms show that SiO₂, Al₂O₃, Fe₂O₃, K₂O, CaO, MgO, SO₃, and Na₂O, were found at almost all locations in variable concentrations. SiO₂ is the most abundant component in the rock samples, which ranges from 55.76 to 63.81 %. This also agrees with the X-ray diffraction results, which indicated that quartz is the most abundant mineral at almost all locations. Furthermore, almost all the minerals detected in the rocks and soil are silicate minerals. In summary, the results range in the following sequence: $SiO_2 > Al_2O_3 >$ $Fe_2O_3 > K_2O > CaO > Na_2O > MgO > SO_3$. The result of silicon oxide is closely followed in abundance by aluminum oxide within the range of 11.06 and 13.15 %; this is also in agreement with the X-ray diffraction result which shows that most of the minerals revealed are of aluminum silicate base. The results also show that the values for iron and potassium oxides are closely comparable and come after aluminum; they are also major components of the minerals detected by XRD in most of the rock samples, though potassium was recorded more than iron. The presence of sodium was observed at only one sample location, BER, where there is the

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presence of albite, a sodium base mineral. This also indicated that BER is the location with highest Na₂O-rich albite. The presence of SO₃ indicated that some sulfide and sulphosalt ore minerals, such as chalcosite Cu₂S, pyrite FeS₂, galena PbS, stibnite Sb₂S₃, and oldhamite CaS, which are susceptible to decomposition under mild conditions [41], are also present in the rock samples. These results agree with the fact that the eight most common elements making up the minerals found in the earth's crust are as follows: oxygen 46.6 %, silicon 27.7 %, aluminum 8.1 %, iron 5.0 %, calcium 3.6 %, sodium 2.8 %, potassium 2.6 % and magnesium 2.1 %, they are really the major elements and make up 98.5 % of the earth, while the remaining 1.5 % are the minor elements. The presence of sulfur oxide is negligible at most locations, indicating that there are no sulfate minerals in all the locations, which also agrees with the results of minerals identified in the area.

The results of the minor elemental oxides (Table 5) provide elements and elemental oxides that were detected in the locations within the study area. TiO_2 (titanium oxide), also known as rutile, is one of the hard rock minerals used for the formulation of text coat paints. It appeared to have remarkable values at all locations. Zinc oxide was detected at only one location, BER, in a minute quantity. This agrees with the XRD results, which show no trace of zinc mineral present at all locations. The results of barium oxide appeared to be comparatively high among the minor elements, especially at BER, but XRD results did not

Location	SiO ₂	Al_2O_3	Fe ₂ O ₃	CaO	MgO	SO ₃	K20	Na ₂ O
BER	61.06	12.68	8.57	2.69	1.30	1.05	6.12	2.40
BNR	55.76	13.15	11.20	5.44	1.90	0.41	7.21	-
BSR	58.59	11.79	9.31	6.37	2.20	0.48	7.75	-
BTR	60.70	12.33	9.72	4.97	0.90	0.51	7.41	-
BWR	63.81	11.06	8.41	3.59	1.50	0.37	7.06	-
Range	55.76	11.06	8.41	2.69	0.90	0.37	6.12	-
	63.81	13.15	11.20	6.37	2.20	1.05	7.75	2.40

Table 4 Percentage (%) composition of major elemental composition in rock samples from Bakin dutse

Table 5 Percentage (%) composition of minor elemental composition in rock samples from Bakin Dutse

Location	TiO ₂	BaO	Zn0	$P_{2}O_{5}$	ZrO ₂	Rb ₂ O	MnO	Cl
BER	0.42	2.06	0.07	0.33	0.62	0.06	0.13	0.14
BNR	2.34	0.15	-	1.35	0.20	0.05	0.18	0.33
BSR	1.41	0.17	-	1.04	0.12	-	0.17	0.32
BTR	1.44	0.08	-	0.44	0.51	0.07	0.17	0.34
BWR	1.94	0.19	-	1.12	-	-	0.13	0.33
Range	0.42	0.08	0.07	0.33	0.12	0.05	0.13	0.14
	2.34	2.06	-	1.35	0.62	0.07	0.18	0.34

Range = Limits between values, - = Not detected.

indicate that there is $BaSO_4$ (barite) in the sample area, which suggests that the deposits may not be high enough for detection by the XRD instrument. Chlorine (in chloride forms) is a typical component of many minerals, they are combined in trace amount and they are typically not involved in their major crystal lattice on the mineral.

The results of the major and minor elemental components of soil samples from Bakin Dutse are presented in Tables 6 and 7, respectively. In similarity to the rock samples, silica (SiO2) appeared as the most abundant mineral available at all the locations, ranging between 46.46 and 55.20 %. Aluminum and Iron oxides appeared competitively at most locations, ranging between 17.43-19.37% and 8.63-18.69%, respectively. Potassium oxide also appeared quantitatively high, a little bit higher than calcium within the range of 5.82 to 7.20%, compared to the results of the rock samples. Sulfur oxide sparingly appeared, just like the results for the rock samples at all the locations. This also indicated that there are traces of some sulfide and sulphosalt ore minerals in all the

locations. Sodium oxide did not appear at all in all the soil samples. The results of the major elemental oxides in soil are in the following sequence; $SiO_2 > Al_2O_3 > Fe_2O_3 > K_2O > CaO > MgO$ > SO_3 . The order of the major elemental oxides in the soil is the same as that obtained for the rock samples in Bakin Dutse. This indicated that the origin of the soil in the area can be linked to the rock. Isaac *et al.* [9] reported that the major oxides in Mambila plateau soil which is also in the northeastern region, follows the order $SiO_2 >$ $Al_2O_3 > TiO_2 > CaO > K_2O$. The study showed TiO_2 as a major oxide and a higher amount of CaO than K_2O , which are in contrast to the results in this study.

The minor elemental composition (Table 7) showed that TiO_2 appeared appreciably within the range 1.32- 3.0 %, followed by P_2O_5 at a range of 0.83-1.94 %. The presence of PbO, CuO, and ZnO were not detected at all in the soil samples. Similar to the rock samples, the presence of chlorine was detected at all the locations sparingly within the range of 0.26-0.33 %. BaO also appeared at all the locations,

Location	SiO ₂	Al ₂ O ₃	Fe ₂ O ₃	CaO	MgO	SO ₃	K20
BES	46.55	17.43	17.04	4.13	2.70	0.65	5.82
BNS	46.46	19.37	18.69	3.09	2.70	0.58	5.91
BSS	47.15	18.94	16.53	3.34	2.50	0.53	5.87
BTS	55.20	17.96	8.63	4.92	2.00	0.61	6.37
BWS	51.46	18.48	12.67	2.71	2.00	0.45	7.20
Range	4.46	17.43	8.63	2.71	2.00	0.45	5.82
	55.20	19.37	18.69	4.92	2.70	0.65	7.20

Table 6 Percentage (%) composition of major elemental composition in soil samples from Bakin Dutse

Range = Limits between values, - = Not detected.

Fable 7 Percentage (%) co	omposition of minor eleme	ntal oxides in soil sam	ples from Bakin Dutse
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Location	TiO ₂	BaO	P_2O_5	ZrO ₂	Rb ₂ O	MnO	Cl
BES	3.00	0.28	1.21	0.24	-	0.27	0.33
BNS	2.91	0.18	0.91	0.25	0.05	0.29	0.33
BSS	2.89	0.33	0.83	0.22	-	0.28	0.33
BTS	1.32	0.06	1.94	0.25	0.08	0.19	0.26
BWS	2.71	0.27	0.86	0.29	0.07	0.21	0.27
Range	1.32 - 3.00	0.06- 0.33	0.83-1.94	0.22-0.29	0.05-0.08	0.19-0.29	0.26- 0.33

Range = Limits between values, - = Not detected.

but in small quantities ranging between 0.06-0.33 %. In summary, the sequence appears as $TiO_2 > P_2O_5 > BaO > ZrO_2 > MnO > Rb_2O.$ Compared to the minor components in Bakin Dutse rocks, PbO, CuO, ZnO, Mn and Cl were not found in the soil. This can be attributed to complete leach-out of these minor components during the weathering of the origin rock to form the soil. Fe_2O_3 was recorded as a major component in this study, while Isaac *et al.* [9], reported Fe_2O_3 as a minor composition of the Mambila plateau soil. This contrast in the amount and order of composition of these compounds can be attributed to different modes of rock weathering. This can be caused by significant difference in mobility of the cations, resulting from the difference in climatic conditions in different locations with the region [42,43].

Conclusion

This study presented the minerals and elemental composition of the rock and soil within the Barkin Dutse hills in Adamawa State, Nigeria. The mineral and elemental compositions of the rock samples from the study area showed the presence (in order of their abundance) of quartz, albite, annite, microcline, orthoclase, phlogopite, oligoclase, chrysotile, and cordierite. The soil sample showed the presence of kaolinite in addition to the minerals recorded in the rocks. The emergence of kaolinite in the soil was attributed to the metamorphosis of the rock's aluminosilicate minerals, such as orthoclase, oligoclase, chrysotile, and cordierite. In general, the results showed similarities to previous studies on the analysis of rocks in the northeastern region of Nigeria, within which the study area falls. The complimentary elemental analyses provided results that are in accordance with the minerals in the study locations. The results obtained in this study will form part of the mineral survey of the northeastern region of Nigeria and make contributions to the geochemical data buildup for the region.

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